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# Fluxes of oxidised and reduced nitrogen above a mixed coniferous forest exposed to various nitrogen emission sources

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Reduced nitrogen was found to be the main contributor to total deposition which was predominantly governed by dry deposition.

#### Abstract

Concentrations of nitrogen gases (NH<sub>3</sub>, NO<sub>2</sub>, NO, HONO and HNO<sub>3</sub>) and particles (pNH<sub>4</sub> and pNO<sub>3</sub>) were measured over a mixed coniferous forest impacted by high nitrogen loads. Nitrogen dioxide (NO<sub>2</sub>) represented the main nitrogen form, followed by nitric oxide (NO) and ammonia (NH<sub>3</sub>). A combination of gradient method (NH<sub>3</sub> and NO<sub>x</sub>) and resistance modelling techniques (HNO<sub>3</sub>, HONO, pNH<sub>4</sub> and pNO<sub>3</sub>) was used to calculate dry deposition of nitrogen compounds. Net flux of NH<sub>3</sub> amounted to -64 ng N m<sup>-2</sup> s<sup>-1</sup> over the measuring period. Net fluxes of NO<sub>x</sub> were upward (8.5 ng N m<sup>-2</sup> s<sup>-1</sup>) with highest emission in the morning. Fluxes of other gases or aerosols substantially contributed to dry deposition. Total nitrogen deposition was estimated at -48 kg N ha<sup>-1</sup> yr<sup>-1</sup> and consisted for almost 80% of NH<sub>x</sub>. Comparison of throughfall nitrogen with total deposition suggested substantial uptake of reduced N ( $\pm 15$  kg N ha<sup>-1</sup> yr<sup>-1</sup>) within the canopy.

Keywords: Ammonia; Gradient method; Resistance model; Oxidised nitrogen; Throughfall

# 1. Introduction

Many forest ecosystems in Flanders are subjected to acidifying and eutrophicating nitrogen inputs. The presence of nitrogen emission sources and the fast removal of gases and aerosols from the atmosphere by enhanced turbulent transfer and the scavenging properties of the canopy lead to the delivery of considerable chemical doses to forests, which may affect their structure and functioning. Simple mass balance calculations for over 1400 forest receptor points in Flanders (1993–1998) indicated that modelled deposition largely exceeded critical loads related to protection of floral

composition, safeguarding balanced nutrient supply and preventing groundwater contamination by nitrate leaching, especially at coniferous receptor sites (Neirynck et al., 2004). Not only does deposition of reduced and oxidised nitrogen contribute to acidification and eutrophication but their presence is also linked to other environmental issues such as photochemical air pollution (Fowler et al., 1998), radiative forcing (Adams et al., 2001) and visibility impairment (Barthelmie and Pryor, 1998).

Monitoring and modelling programmes have been launched to improve understanding and measurement of surface exchange of trace gases and particles (Erisman et al., 2005). Results are needed to allow the evaluation of abatement strategies and calculation of regional deposition budgets of atmospherically derived nitrogen (Butler et al., 2005; Fowler et al., 2005; Sutton et al., 2003).

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As an alternative to expensive and time-consuming micrometeorological measurements and resistance modelling, throughfall measurements are often used to determine the nitrogen flux in forests. This form of monitoring is cost-effective and was currently being carried out in more than 500 level II-plots across Europe (Lorenz et al., 2006). A significant drawback, however, is that throughfall underestimates total nitrogen flux, since a proportion of nitrogen may be taken up by the canopy or consumed by canopy-dwelling micro organisms (Lovett and Lindberg, 1993). Moreover, throughfall does not include the dry deposition flux of gases and fine aerosols directly to the soil (Lindberg et al., 1986).

This paper focuses on the presence and exchange properties of different oxidised and reduced nitrogen forms above mixed coniferous forest, exposed to nearby nitrogen emission sources. Fluxes of these compounds are obtained through a combination of gradient measurements and resistance modelling in order to obtain an insight into their relative contribution to the dry deposition flux. Additionally, dry and wet deposition fluxes are compared to throughfall nitrogen data in the forest in order to gain a clearer understanding of how throughfall samplers underestimate the total nitrogen flux.

#### 2. Material and methods

### 2.1. Site description

The measurements are conducted over a mixed coniferous/deciduous forest located in the Campine region of Flanders (Belgium, 51°18′ N, 4°31′ E) approximately 15 km northeast of Antwerp. The forest encompasses over 300 ha, and is heterogeneous but of even height. The landscape is a coastal plain, with a gentle (0.3%) slope. The measurement site in the forest concerns a 2 ha large Scots pine stand (*Pinus sylvestris* L.) planted in 1929. Other Scots pine stands surround (ca. 150–300 m) the measurement site, with patches of deciduous trees found further away. It is bordered to the north and west by residential areas of the town of Brasschaat at a distance of ca. 500 m (Fig. 1), and to the south and east the forest extends over 2 km before turning into rural, partially forested terrain.

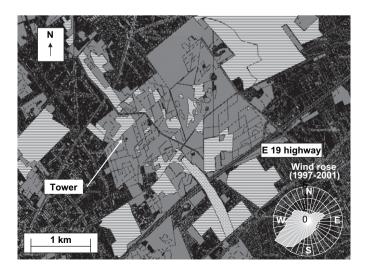


Fig. 1. Location of the measurement tower in the experimental forest site (grey: forest, black: residential areas, waves: water pools, horizontal bands: low vegetation types such as meadows, clearcuts or moorlands).

The forest is exposed to different pollution climates. Southerly to westerly winds bring  $SO_2$  and  $NO_x$ -bearing air masses originating from either the petrochemical industry in the harbour of Antwerp (15 km to the west) or from E19 highway traffic (2 km to the south), suburban traffic and heating from contiguous buildings. Ammonia emission — from cattle stables and manure spreading — originates from rural communities located approximately 10 km north and east and the concerning grid cells have an emission flux density ranging between 4 and 11 t N km $^{-2}$  yr $^{-1}$ . Manure spreading also takes place within 5 km of the forest.

# 2.2. Meteorological measurements

A self-supporting welded square scaffolding tower was erected in the measuring site to a height of 40 m, with a 9 m<sup>2</sup> ground area and platforms at 9, 15, 18, 23, 31 and 39 m, next to a level II observation plot of the European ICP-Forests network (EC-UN/ECE, 1996), which also features in the CARBO-EUROPE IP research network (http://www.carboeurope.org). Meteorological data included vertical profiles of air temperature and humidity (psychrometer, Didcot DTS-5A, UK) at 2, 10, 24, 32 and 40 m and wind speed (cup anemometer, Didcot DWR-205G, UK) at 24, 32 and 40 m. At the top of the tower, measurements were made of down-welling shortwave radiation (pyranometer, Kipp and Zonen CM6B, The Netherlands), photosynthetic photon flux density (PPFD quantum sensor, JYP 1000, JDEC, France) and precipitation (tipping bucket, Didcot DRG-51, UK). A leaf wetness sensor (237F, Campbell, UK) was mounted on a boom at 19 m. All meteorological sensors were sampled at 0.1 Hz and stored as half hour means on a data logger (Campbell CR10, UK). A 3D sonic anemometer (model SOLENT 1012R2, Gill Instruments, UK) was deployed on a mast above the tower at 41 m.

#### 2.3. Measurements of N compounds

#### 2.3.1. Sampling and analysis of N-containing gases and particles in air

Measurements of gaseous  $NO_x$  (chemiluminescence, Ecophysics 700 AL, Switzerland, detection limit 1 ppb) and  $NH_3$  (Conductivity, AMANDA, ECN, The Netherlands, detection limit 0.1  $\mu$ g m<sup>-3</sup>) were conducted at three levels above the canopy (23, 31 and 39 m) every half hour. For  $NO_x$ , air was drawn through heated Teflon tubing (35 °C) to the measuring unit at ground level. For  $NH_3$ , wet rotating annular denuders (Wyers et al., 1993) were deployed at the platforms to trap gaseous  $NH_3$  into an acid solution ( $NaHSO_4 \cdot H_2O$  0.5 g/l) rotating in the denuders. The solution was transported from the denuders to a heated detection cell at the 23 m level platform. Ammonia gradients were only available for the period June 1999—November 2001. Gradient measurements of  $NO_3$ , discussed in this paper, applied to the same period.

Gases of nitrous acid (HONO), nitric acid (HNO<sub>3</sub>) and fine particles of ammonium (pNH<sub>4</sub>) and nitrate (pNO<sub>3</sub>) were measured on a daily basis, in three separate sampling campaigns, during the time when the AMANDA analyser was operational. Daily measurements of NH<sub>3</sub> and pNH<sub>4</sub> were conducted by the Flemish Institute for Technological Research (Berghmans et al., 2001) between September 1999 and July 2000<sup>1</sup> using a glass honeycomb denuder—filter pack system (Koutrakis et al., 1993). The sampling cartridges consisted of a PTFE-coated PM<sub>2.5</sub> keyhole impactor, a glass transition section, two honeycomb denuders, a spacer, and a filter pack. The cartridges were attached to a sampling manifold and a sample changer deployed at the 24 m platform. This device controlled the flow through seven ports with attached cartridges, which were sampled consecutively starting from midnight onwards. Air was drawn at a rate of  $10 \, \text{L} \, \text{min}^{-1}$  at which the impactor had a 50% cut point at 2.5  $\, \text{\mu m}$ .

Denuders were coated with a methanol solution containing 1% glycerol and 2% citric acid to remove NH<sub>3</sub>. Aerosols were caught on the filter pack deployed behind the denuder, which consisted of a Teflon filter (Teflon 47 mm,  $2 \mu m$ , Pall—Gelman P/N R2PJ047) for particle collection and two additional quartz fibre filters (QF20 47 mm, Schleicher & Schuell Ref. No. 10373219) impregnated with a methanol solution of 2% citric acid and 2% glycerol to

 $<sup>^{\</sup>rm 1}$  No measurements taken in January 2000 and additional daily measurements of NH  $_{\rm 3}$  were taken in September/October 2000.

collect revolatilized NH<sub>3</sub> from the Teflon filters. After being coated, the denuders were immediately dried with clean dry air and capped to prevent contamination by NH<sub>3</sub>. The cartridges were assembled in the laboratory and checked for leaks, before being transported to the sampling site.

The same setting was used to perform daily samplings of pNO<sub>3</sub>, HONO and HNO<sub>3</sub>, conducted in January–March 2001 (winter campaign  $NO_y$  2001). Gases were collected on coated honeycomb denuders (coating solution of 50/50 methanol/ultrapure water with 1% glycerol and 1% Na<sub>2</sub>CO<sub>3</sub>). A nylon membrane filter (Nylasorb 47 mm 1.0  $\mu$ m, Pall–Gelman P/N 66509) was used for collecting particles. Quartz fibre filters (QF20 47 mm, Schleicher & Schuell Ref. No. 10373219) with coating solutions of 2% glycerol and 1% Na<sub>2</sub>CO<sub>3</sub> were positioned behind the nylon filter for absorbing evaporated acid gases produced from reactions at the nylon filter surface.

After sampling, the cartridges were transported to laboratory and dissembled under acid-aerosol-free conditions. For the NH3 measurements, denuders were extracted with 10 mL of ultrapure water. The extraction solution for the HONO and HNO<sub>3</sub> denuder measurements included 0.239 g/L Na<sub>2</sub>CO<sub>3</sub> and 0.235 g/L NaHCO<sub>3</sub> in deionised water. For the filter samples (pNH<sub>4</sub>, pNO<sub>3</sub>) an extraction volume of 20 mL was used. The samples were stored at 4 °C and usually analysed within 1 week. All sampling preparations were carried out in an acidic gas-free laminar flow box; the water met Millipore quality (conductivity  $< 0.05 \,\mu\text{S cm}^{-1}$ ). The ionic concentrations of the aqueous extracts were determined using ion chromatography (Dionex 120-SE, separation columns AS14A for anions and CS12A for cations) under standard conditions. Accuracy was checked by certified standards (Spex IC Instrument Check Standard 6 (cat. No. ICMIX6-100) for anions and (Spex IC Instrument Check Standard 4 (cat. No. ICMIX4-100) for cations. The overall accuracy in IC analysis was around 5% for all analytes. Field blanks were taken every week during measurements. The procedural blanks were collected in the field with a passive exposure to air. Blank levels, typically  $\sim 1.5 \,\mu g$  for NH<sub>3</sub>,  $\sim 2.3 \,\mu g$  for HNO<sub>3</sub> and  $\sim 1.7 \,\mu g$  for HONO, were subtracted from the detected amounts. The minimum detectable limits (defined as the standard deviation of the field blanks and based on an assumed mean air volume of 14.6 m<sup>3</sup>) were (in  $\mu$ g m<sup>-3</sup>) 0.07 for NH<sub>3</sub>, 0.06 for HONO, 0.08 for HNO<sub>3</sub>, 0.06 for pNH<sub>4</sub><sup>+</sup>, and 0.09 for pNO<sub>3</sub><sup>-</sup>. For the measurement of HONO and HNO<sub>3</sub>-gas sampling artifacts of 3-4% for HNO<sub>3</sub> and 5-10% for HONO have been reported (Bai et al., 2003; Koutrakis et al., 1988). At temperatures >28 °C, HNO<sub>3</sub> readily absorbs and desorbs on surfaces due to interference of NO<sub>2</sub>. Mean temperatures during the winter campaign NO<sub>y</sub> 2001 were considerably lower (5 °C). The determination of N-species on uncoated quartz fibre filters can suffer from sampling artifacts like evaporation of semivolatile ammonium nitrate producing equimolar amounts of HNO<sub>3</sub> and NH<sub>3</sub> (Harrison and Pio, 1983) or retention of ambient HNO3 on the filter (Appel et al., 1984). The use of denuder-filter systems is employed to prevent these artifacts. During both sampling campaigns pre-extracted quartz fibre filters were used.

Additional measurements of oxidised nitrogen on a daily basis were made by the University of Antwerp (Eyckmans et al., 2001) during the months of July, September and October 2000 (summer/early autumn campaign NO<sub>y</sub> 2000). The sampling was made using a dual channel sequential fine particle sampler (URG 2000-01 K, United Research Glassware, NC, USA). Coarse particles above 10 µm diameter were separated at the air inlets. Air masses were drawn at a flow rate of 10 L min<sup>-1</sup> through annular denuders in front of which a cyclone removed the particle fraction above 2.5 µm. For acid gases, a coating of methanol/ultrapure water (50/50), in which 1% glycerol and 1% Na<sub>2</sub>CO<sub>3</sub> were dissolved, was applied to three concentric annular denuders. Teflon filter-holders placed behind the denuders were cleansed on a regular basis and loaded with Whatman filters (GF/F, 47 mm), and denuder sets were replaced every 4 days. Exposure duration and time of denuder change (midnight) were programmed through an electronic switching system.

Denuder coating was extracted with  $2\times 5$  mL Milli-Q (Millipore) water and analysed using ion chromatography (Dionex DX 120). The filters themselves were put in a plastic jar along with 10 mL Milli-Q water and extracted for 20 min in an ultrasonic bath. Afterwards, the solution was filtered through a 0.22  $\mu$ m pore Millex filter (Millipore) and analysed using IC.

# 2.3.2. Throughfall water

Throughfall (TF) was sampled using 10 systematically distributed bulk collectors in the adjacent 0.25 ha large level II observation plot. They

consisted of a polyethylene funnel (14 cm in diameter) placed at a standard height of 1 m, which was connected to a subterranean 2 L polyethylene bottle. A nylon mesh (1 mm) was placed in the funnel to avoid contamination by large particles. Samples from all collectors were pooled together for every sampling event, which took place 24 times a year. When insufficient rainfall was registered, sampling was postponed without replacing the sampling equipment. Bulk precipitation was collected using four bulk collectors placed at an adjacent pasture (about 250 m away from the tower location) with the same sampling frequency. Fractions of  $NH_4^+$  and  $NO_3^-$  were analysed using ion chromatography (Dionex DX-100). Conversion factors from bulk deposition to wet deposition samplers were derived from a comparison exercise between bulk collectors and a wet-only sampler at another level II site near to Ghent (50°58′ N, 3°49′ E; see Staelens et al., 2005) and were estimated to be 85% and 79% for  $NH_4^+$  and  $NO_3^-$ , respectively.

#### 2.4. Calculation of nitrogen fluxes

Both gradient measurements and resistance models were applied to estimate ecosystem input. For the gases measured every half hour (NH<sub>3</sub> and NO<sub>x</sub>), gradients from the 23 and 39 m interval (lower and upper profile height) were used to calculate fluxes applying the gradient method (Dyer and Hicks, 1970):

$$F = -K \frac{\partial c}{\partial \tau} \tag{1}$$

where F is the flux (in  $\mu$ g m<sup>-2</sup> s<sup>-1</sup>, defined positive upward), c is the concentration ( $\mu$ g m<sup>-3</sup>), z is the height (in m) and K is the turbulent diffusivity (in m<sup>2</sup> s<sup>-1</sup>), calculated as:

$$K = \frac{k(z-d)u_*}{\phi} \tag{2}$$

In this formula k (the von Karman constant) is 0.4, z is the geometric mean of the measurement heights (29.9 m), d is the zero plane displacement (=19.2 m) inferred from wind profile measurements, and  $u_*$  is the friction velocity determined as the (negative) square root of the kinematic momentum flux measured by eddy covariance. In order to account for stability effects, the universal flux—profile relationships for heat transfer  $(\phi_h)$  were applied (Dyer and Hicks, 1970). Because the concentration measurements were made in the roughness sublayer, turbulent diffusivities estimated by Eq. (2) were corrected by a factor  $(\alpha, \text{here} = 0.87)$  to allow for wake turbulence generated above the canopy (Bosveld, 1991):

$$\phi_{\rm h} = \begin{cases} L \le 0.... \alpha \times \left(1 - 16\frac{(z-d)}{L}\right)^{-1/2} \\ L > 0.... \alpha + 5\frac{(z-d)}{L} \end{cases}$$

$$(3)$$

where L is the Monin-Obukhov length and (z-d)/L the dimensionless stability height.

Data were screened to exclude measurement problems and conditions invalidating the use of flux-gradient theory. Friction velocities below 0.1 m s<sup>-1</sup> were rejected because of probable invalid flux—profile relationships. A footprint analysis, carried out by Göckede et al. (2005) based upon source weight functions for all stratification regimes, revealed that the forest land use type contributed on average about 80% to each eddy covariance flux measurement at 41 m height. Analysis of horizontal structures in the raw turbulent data disclosed no disturbing influence of the surrounding landscape on the eddy covariance measurements. Therefore, no further rejection criteria were included to avoid potential perturbation effects due to limited fetch in NW direction. To avoid non-stationarity, data were excluded for which concentrations' changes lead to half hour changes in deposition velocity  $v_d$  exceeding 0.01 m s<sup>-1</sup>,  $(|z/c \times (dc/dt)| < 0.01 \text{ m s}^{-1})$ . Gradients were checked for systematic bias between the two heights during episodes of high turbulence ( $u*>1.5 \text{ m s}^{-1}$ ) and strong rain (>4 mm h<sup>-1</sup>) for which gradients were expected to drop to zero. Unequal flow rates due to denuder malfunction or blocked sample lines (presence of soot or pollen) lead to biased gradients (time shift between denuder readings), which were omitted from the NH<sub>3</sub> data set. In order to reduce the relative errors, concentrations below 0.1 µg m<sup>-3</sup> (in case of NH<sub>3</sub>) or 1 ppb

(in case of  $NO_x$ ) were excluded. Outliers in the data were removed, rejecting any deposition velocity exceeding  $2/R_a$ .

For NO and NO<sub>2</sub>, flux divergence could possibly occur due to chemical reactions within the O<sub>3</sub>/NO/NO<sub>2</sub> triad (Vilà-Guerau de Arellano and Duynkerke, 1992). Time scales for chemical reactions ( $\tau_c$ ) were compared with time scales of turbulent transfer ( $\tau_t$ ) in order to ensure that the constant flux assumption was not invalidated. The chemical reaction timescale  $\tau_c$  was calculated according to Lenschow (1982), whereas  $\tau_t$  was approximated according to Ammann (1999).

Fluxes of other gaseous compounds HONO and HNO<sub>3</sub>, sampled on daily basis, were modelled. It was assumed that the gases were continuously deposited and no surface concentration existed:

$$F = -\nu_{\rm d}(z)c(z) \tag{4}$$

where  $v_d(z)$  is the deposition velocity at height z (m s<sup>-1</sup>), and c(z) is the concentration at reference height z.

Deposition velocity was inferred from measured meteorological variables using a resistance analogy in which  $\nu_d$  is calculated as the inverse of three resistances (Hicks et al., 1987):

$$\nu_{\rm d}(z) = \frac{1}{R_{\rm a}(z) + R_{\rm b} + R_{\rm c}} \tag{5}$$

where  $R_a(z)$  is the aerodynamic resistance,  $R_b$  is the laminar sublayer resistance, and  $R_c$  is the canopy resistance.

Aerodynamic resistance  $R_a(z)$  was calculated according to Garland (1978) and was the same for all gases:

$$R_{\rm a}(z) = \frac{1}{ku_*} \left[ \ln \left( \frac{z - d}{z_0} \right) - \psi_{\rm h} \left( \frac{z - d}{L} \right) + \psi_{\rm h} \left( \frac{z_0}{L} \right) \right] \tag{6}$$

where  $z_0$  is the roughness length (1.4 m), and  $\psi_h((z-d)/L)$  is the integrated stability correction for heat, estimated following Beljaars and Holtslag (1990).

 $R_{\rm b}$  was species-dependent and estimated using semi-empirical relationships presented by Hicks et al. (1987) with Schmidt and Prandtl number corrections:

$$R_{\rm b} = \frac{2}{ku_*} \left(\frac{\rm Sc}{\rm Pr}\right)^{2/3} \tag{7}$$

The canopy resistance  $(R_c)$  of gases consists of stomatal resistance  $(R_s)$ , soil  $(R_{soil})$  and in-canopy aerodynamic resistance  $(R_{inc})$ , together with cuticular resistance  $(R_w)$  acting in parallel:

$$R_{\rm c} = \left[ \frac{1}{R_{\rm s}} + \frac{1}{R_{\rm inc} + R_{\rm soil}} + \frac{1}{R_{\rm w}} \right]^{-1} \tag{8}$$

Terms in this equation were estimated for different gases using parameterisations mentioned in Erisman et al. (1994). Stomatal resistance ( $R_{\rm s}$ ) was calculated using the parameterisations by Wesely (1989) based on the method by Baldocchi et al. (1987). This parametrisation only needed data for global radiation and surface temperature and was corrected for the different diffusion coefficient of the gas involved compared to water vapour. Formulation for  $R_{\rm inc}$  was modelled by Erisman et al. (1994). For HNO<sub>3</sub> it was assumed that it was deposited along the leaf surface with negligible  $R_{\rm w}$ . Also for the soil a negligible  $R_{\rm soil}$  was assumed. For HONO, data on  $R_{\rm w}$  and  $R_{\rm soil}$  were lacking and instead parameterisations for SO<sub>2</sub> deposition on wet surfaces were used because of a similar chemical propensity (Erisman et al., 1994).

Deposition velocity ( $\nu_d$ ) for particles (pNH<sub>4</sub> and pNO<sub>3</sub>) was calculated using the modified model of Slinn (1982):

$$\nu_{\rm d} = \nu_{\rm g} + \left[ R_{\rm a}(50) + \nu_{\rm ds}^{-1} \right]^{-1} \tag{9}$$

where  $\nu_{\rm g}$  is the gravitational settling velocity of particles and  $\nu_{\rm ds}$  is the surface deposition velocity, and  $R_{\rm a}(50)$  is the aerodynamic resistance at 50 m height. Due to only small-sized particles ( $\nu_{\rm g}\approx 0$ ) having been collected, Eq. (9) was reduced to a turbulent contribution and a surface deposition velocity. Particle deposition was modelled following the assumption of similarity to momentum deposition (Chamberlain, 1966; Slinn, 1982):

$$\nu_{\rm ds} = E \frac{u_*^2}{u_{\rm b}} \tag{10}$$

where  $u_*$  is the friction velocity,  $u_h$  is wind speed at the top of the canopy, and E is the total collection efficiency with which particles are captured by the canopy. The collection efficiency for pNO<sub>3</sub> and pNH<sub>4</sub> was calculated using parameterisations presented by Ruijgrok et al. (1997) made for coniferous forest. The relationships used for calculating E took into account different responses among the depositing components for conditions of  $u_*$ , leaf wetness (no particle rebound for wet surface) and ambient relative humidity (hygroscopic growth).

#### 2.5. Data handling and statistics

Given the problems in handling the AMANDA analyser in harsh winter conditions and the removal of biased gradients, the data coverage was 33% during the operation period (June 1999-November 2001). In order to obtain complete budgets for NH3, gap filling was performed using a single-hiddenlayer feed-forward neural network (Saxén and Saxén, 1995). Given the importance of canopy wetness in ammonia deposition patterns (Neirynck et al., 2005), the half hourly data set was subdivided into different canopy wetness categories: rainy conditions (rainfall measured by pluviometer) and non-rainy conditions (no rainfall measured by pluviometer). The latter were further subdivided into three classes depending on the grid wetness of the leaf wetness sensor: 0% (dry); 0% < wetness < 100% (wet); 100% (saturated). The neural network (NN) training and subsequent gap filling was carried out separately for each wetness category. The complete data set from each category was randomly divided in two even-sized sets. Neural networks were trained with relevant variables (such as NH<sub>3</sub>,  $u_*$ , (z-d)/L, temperature, ...) from one data set (training data set), evaluated with the other (test data set) and then used to fill missing values of NH3 flux. When no NH3 measurements from the monitor (single height) were available, daily measurements from honeycomb denuders (NHx campaign) were used as input to fill missing fluxes. Networks were also trained without information regarding NH<sub>3</sub> or molar NH<sub>3</sub>/SO<sub>2</sub> ratio. When data on atmospheric turbulence were lacking, no NN were trained and no gap filling was done. Fluxes for which no robust calculation was possible because concentrations fell below the detection limit ( $c < 0.1 \, \mu \mathrm{g \, m}^{-3}$ ) or low turbulence ( $u_* < 0.1 \text{ m s}^{-1}$ ) prevailed, were set to zero given the expectation of negligible fluxes.

Given the seasonal differences in concentrations and fluxes, data set was subdivided into a summer season (April—September) and a winter period (October—March).

# 3. Results

# 3.1. Composition and variability of nitrogen in atmosphere

Nitrogen dioxide ( $NO_2$ ) was the prevailing nitrogen compound in the lower troposphere at the site (Table 1). Its average concentration of  $8.7 \,\mu g \, N \, m^{-3}$  was two to three times higher compared to nitric oxide (NO) and ammonia ( $NH_3$ ). Nitrogen oxides were higher in concentration than  $NH_3$  during the winter (Fig. 2A). During summer, levels of  $NH_3$  strongly increased and even approached the declining  $NO_2$  levels (Fig. 2B).

Nitrogen oxides exhibited pronounced diurnal patterns that featured a strong morning peak for NO (Fig. 2A and B). The NO peak occurred earlier in the morning during the summer and was lower in magnitude. Peaking occurred approximately 1 h after sunrise time. The diurnal pattern of NH<sub>3</sub> was flattened during the winter period but more susceptible to atmospheric mixture during the summer.

Table 1 Statistical parameters of nitrogen compounds (NH<sub>3</sub>, NO and NO<sub>2</sub>, pNH<sub>4</sub>, HONO, HNO<sub>3</sub>, pNO<sub>3</sub>) measured at 0.5 h or 24 h time steps above the canopy for measurements conducted between July 1999 and November 2001 (in  $\mu$ g N m<sup>-3</sup>)

Compound	Period	n	Arith av	σ	median	min	max	Geom. Av	Geom. σ
0.5 h sampling									
$NH_3$	Jul 99-Nov 01	15,047	3.0	5.6	1.3	0.0	102.5	1.2	3.5
NO	Jul 99-Nov 01	39,226	3.9	12.5	0.0	0.0	291.7	1.3	2.0
$NO_2$	Jul 99-Nov 01	39,228	8.7	6.2	7.5	0.0	55.4	6.9	1.3
24 h sampling									
$pNH_4$	Sep 99-Jun 00	225	2.0	1.6	1.5	0.2	8.8	1.5	1.7
NH <sub>3</sub>	Sep 99-Oct 00	299	2.4	2.6	1.5	0.1	17.5	1.6	1.9
HONO	Jul, Sep, Oct 00	72	0.5	0.4	0.4	0.0	1.4	0.4	0.8
	Jan, Feb, Mar 01	85	0.5	0.5	0.4	0.0	2.6	0.4	0.7
$HNO_3$	Jul, Sep, Oct 00	72	0.15	0.15	0.12	0.00	1.13	0.13	0.4
	Jan, Feb, Mar 01	85	0.05	0.06	0.02	0.02	0.42	0.03	0.5
$pNO_3$	Jul, Sep, Oct 00	67	0.5	0.3	0.4	0.0	1.8	0.4	0.5
-	Jan, Feb, Mar 01	85	1.2	0.9	1.1	0.0	3.8	0.9	0.5

Particles of NH<sub>4</sub>, which were measured during eight nonconsecutive months on a daily basis (Table 1), were lower in arithmetic mean concentration than NH<sub>3</sub> captured by denuders but median and geometric mean were similar to those of NH<sub>3</sub>. The two compounds were significantly correlated ( $R^2 = 0.41$ , n = 226) and exhibited highest levels during summer time. Particles of NO<sub>3</sub> were particularly abundant during the winter campaign, in contrast to HNO<sub>3</sub>, which only appeared in substantial quantities during summer/early autumn. During both campaigns, levels of HONO were more elevated than those of HNO<sub>3</sub>.

Levels of measured nitrogen compounds were strongly dependent on wind direction (Fig. 3A and B), determined by nearby emission sources. High levels of NO, NO<sub>2</sub> and HONO originated from the W-ESE sector. Nitric oxide reached its highest levels in the SSW-S sector, whereas HONO peaked in the ESE or S sector. Nitrous acid was correlated with NO and NO<sub>2</sub> ( $R^2 = 0.42$ , n = 127). Nitric acid (wind sector dependence not shown) was uncorrelated to these possible precursors and other nitrogen compounds. North-easterly winds brought NH<sub>x</sub>-enriched air masses to the site, although pNH<sub>4</sub> seemed to be associated with a larger range of wind directions than NH<sub>3</sub>. Nitrate particles, possibly pertaining to ammonium nitrate-bearing particles, peaked when winds blew from the ENE sector.

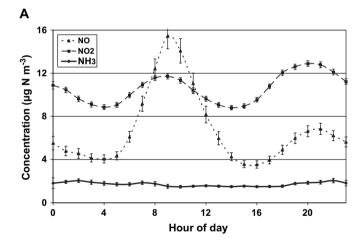
# 3.2. Atmospheric nitrogen fluxes

# 3.2.1. Characteristics of dry deposition fluxes

After applying the above-mentioned rejection criteria, 8824 half hourly flux values were retained with an average net NH<sub>3</sub> flux of  $-75 \pm 145$  ng N m<sup>-2</sup> s<sup>-1</sup> with a corresponding  $\nu_{\rm d}$  of  $3.0 \pm 4.6$  cm s<sup>-1</sup> (Neirynck et al., 2005; Table 2). Net NH<sub>3</sub> flux measured during the summer period was three times higher compared to winter period flux. This was rather due to lower NH<sub>3</sub> levels during winter since average  $\nu_{\rm d}$  differed only slightly across season (Table 3). The small differences in  $\nu_{\rm d}$  could be explained by the fact that daily atmospheric resistances ( $R_{\rm a} + R_{\rm b}$ ) and canopy resistance ( $R_{\rm c}$ ) varied only slightly

among season. At nighttime, however, summer  $R_{\rm a}+R_{\rm b}$  and  $R_{\rm c}$  were almost two times higher compared to winter season values which resulted in a lower nighttime summer  $\nu_{\rm d}$  of  $1.7~{\rm cm~s}^{-1}$ .

Neural networks (NN) were trained with measured NH<sub>3</sub> flux data and used to predict missing values for different



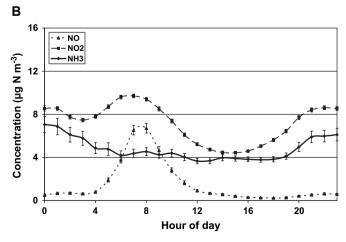
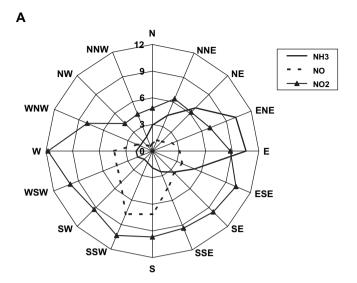


Fig. 2. Averaged diurnal course of NO,  $NO_2$  and  $NH_3$  concentrations (standard error shown by bars) for winter (A) and summer (B) at a height of 40 m.



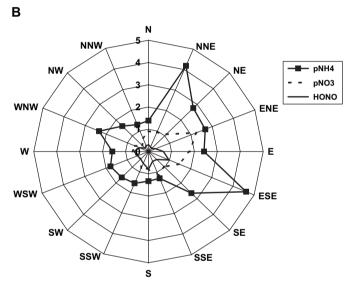


Fig. 3. Wind sector dependencies of average concentration of NH<sub>3</sub>, NO, NO<sub>2</sub> (A) and HONO, pNO<sub>3</sub> and pNH<sub>4</sub> (B) for the Brasschaat forest site during the period June 1999—November 2001 (in  $\mu$ g N m<sup>-3</sup>) (HNO<sub>3</sub> is not shown because of low concentrations).

canopy wetness categories (definition see above). Explanatory variables in the training set were found to be  $u_*$ , (z-d)/L, NH<sub>3</sub>, NH<sub>3</sub>/SO<sub>2</sub>, temperature (T), relative humidity (RH), photosynthetic photon flux density (PPFD) and wind direction. When trained NN performance was evaluated with test data sets (Table 4), best agreement between measurements and NN values was obtained when canopy was dry  $(R^2 = 0.59;$  Fig. 4). Predictions made by NN were worst during rainy conditions; a departure >25% from measured mean and median flux was found and  $R^2$  was only 30%.

Gap filling of NH<sub>3</sub> fluxes assuming null NH<sub>3</sub> fluxes for low turbulence or low concentration data resulted in an increase of data coverage from 20% (n = 8824) to 33% (n = 14.560). When also NN predicted data were used, data coverage rose to 87% (n = 37,123) (Fig. 5). This was especially helpful for achieving a better flux estimate for winter conditions during which measuring equipment was difficult to handle or not operational which was the case in winter 2000-2001. An average net NH<sub>3</sub> flux of  $-64 \pm 120$  ng N m<sup>-2</sup> s<sup>-1</sup> was obtained for the entire measuring period (Table 2). Same conclusions drawn for the measured fluxes applied also for the completed flux data set but fluxes were about 10-15% lower compared to the measured fluxes and the average  $v_{\rm d}$  decreased to  $2.8 \pm 5.1~{\rm cm~s}^{-1}$ . Measured fluxes and fluxes including gap filled data featured a different diurnal pattern during both winter and summer (Fig. 6A). Measured fluxes were always higher compared to the gap filled data set except at noon, during which fluxes dropped as a result of increased emission or reduced uptake at the canopy.

For NO<sub>x</sub>, the presence of flux divergence above the canopy was tested before the gradient method could be applied. An analysis of the timescale ratios  $\tau_c/\tau_t$  revealed that for 87% of the observations, turbulence time scales ( $\tau_t$ ) were more than one order of magnitude smaller than chemical reaction time scales ( $\tau_c$ ). The observations during which response time of diffusive transfer was in the same order of magnitude as chemical reactions coincided with calm, stable (nighttime) conditions. These observations ( $u_* < 0.1 \text{ m s}^{-1}$ ) were, however, rejected from the data set because fluxes were uncertain to calculate. Slow chemistry between the reference height and the surface could therefore be assumed and a direct flux calculation of the NO/NO<sub>2</sub>/O<sub>3</sub> triad by the gradient method was

Table 2 Total and seasonal averages and standard deviations of flux characteristics of  $NH_3$  (measured and including gap filled data) and  $NO_x$  measured on a half hourly basis during the period July 1999—November 2001

-		Method	n	$F (\text{ng N m}^{-2} \text{s}^{-1})$	$\sigma F (\text{ng N m}^{-2} \text{ s}^{-1})$	$v_{\rm d}~({\rm cm~s}^{-1})$	$\sigma v_{\rm d} \ ({\rm cm \ s^{-1}})$
NH <sub>3</sub>	Total	Gradient meas.	8824	-75	145	3.0	4.6
	Winter <sup>b</sup>		3729	-36	69	3.2	4.8
	Summer		5095	-104	176	2.8	4.5
$NH_3$	Total	Gradient + gap filling <sup>a</sup>	37,123	-64	120	2.8	5.1
	Winter		15,480	-31	56	2.9	5.1
	Summer		18,077	-93	150	2.7	5.0
$NO_x$	Total	Gradient meas.	26,603	+8.5	259	-0.04	3.1
	Winter		11,905	+9.4	282	0.00	2.8
	Summer		14,698	+7.8	239	-0.07	3.4

<sup>&</sup>lt;sup>a</sup> Including gap filled data obtained by neural network and previously rejected data for which NH<sub>3</sub> levels <0.1  $\mu$ g m<sup>-3</sup> and  $u_*$  < 0.1 m s<sup>-1</sup> which were assumed to be zero.

<sup>&</sup>lt;sup>b</sup> Winter period: October-March; Summer: April-September.

Table 3
Characteristics of NH<sub>3</sub> flux measurements during summer and winter period

	-					
	Summer			Winter		
	Daytime	Nighttime	Whole day	Daytime	Nighttime	Whole day
n	3238	1857	5095	1496	2233	3729
$NH_3 (\mu g N m^{-3})$	4.0	4.3	4.1	1.5	1.6	1.6
Flux (ng N m $^{-2}$ s $^{-1}$ )	-132	-54	-104	-40	-34	-36
$v_{\rm d}~({\rm cm~s}^{-1})$	3.4	1.7	2.8	3.6	3.0	3.2
$R_{\rm a} + R_{\rm b}  (\mathrm{s  m}^{-1})$	23	53	34	23	34	30
$R_{\rm c}50~({\rm s~m}^{-1})^{\rm a}$	9.1	35.3	15.9	13.1	19.8	17.4

<sup>&</sup>lt;sup>a</sup>  $R_c$ 50, median from  $1/\nu_d - R_a - R_b$  after omitting upward fluxes.

tenable. It was decided take to  $NO_x$  ( $NO + NO_2$ ) gradients to calculate the total  $NO_x$  flux because it was a more robust measurement based estimate of  $NO_x$  exchange (sum stays conserved) and less  $NO_x$  fluxes were omitted because  $NO_x$  levels were seldom below the detection limit (1 ppb) compared to a separate analysis.

Net NO<sub>x</sub> flux was upward (8.5 ng N m<sup>-2</sup> s<sup>-1</sup>) with a netemission velocity  $\nu_{\rm d}$  of -0.04 cm s<sup>-1</sup> (Table 2). No marked differences in flux magnitude between summer and winter period could be detected. Highest emissions up to 40 ng N m<sup>-2</sup> s<sup>-1</sup> were observed in the morning (Fig. 6B) but NO<sub>x</sub> exchange tended to net-deposition at noon.

Other N compounds, measured every 24 h, contributed significantly to the dry flux of nitrogen (Table 5). The flux of pNH<sub>4</sub> amounted to -20 ng N m<sup>-2</sup> s<sup>-1</sup>, which comprised about 30% of the net NH<sub>3</sub> flux. The dry deposition flux of the remaining oxidised nitrogen compounds amounted to -20 and -27 ng N m<sup>-2</sup> s<sup>-1</sup> for the summer and winter campaigns, respectively (Table 5). The flux of HONO amounted to about -9 ng N m<sup>-2</sup> s<sup>-1</sup>, irrespective of season. Contributions of pNO<sub>3</sub> and HNO<sub>3</sub> presented important seasonal changes. During the winter campaign deposition of pNO<sub>3</sub> was three times higher than the summer flux because of both higher  $\nu_{\rm d}$  (due to higher relative humidity and leaf wetness during winter) and measured concentrations. The opposite was the case for HNO<sub>3</sub>, the flux being more important during the summer period when its concentration was substantially higher.

# 3.2.2. Total nitrogen deposition versus throughfall

Total nitrogen deposition (TD), calculated as the sum of wet deposition (WD) and dry deposition (DD), was estimated at  $-48 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  (Table 6). Wet deposition of NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup>, derived from bulk deposition, amounted to -17.5 kg and comprised 36% of TD. Wet deposition consisted mainly

Table 4 Comparison of measured fluxes from test files with their output results from trained neural networks (NN) for different canopy wetness categories (in ng N m $^{-2}$  s $^{-1}$ )

Canopy wetness	n	Mean fl	uxes	Median	fluxes	Performance
		Meas.	NN	Meas.	NN	$R^2$
Dry	1547	111	108	34	38	0.59
Wet	622	54	63	25	25	0.40
Saturated	1135	40	42	14	15	0.54
Rainy	198	38	48	23	18	0.28

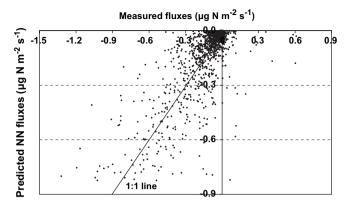


Fig. 4. Measured fluxes from test data set versus fluxes predicted by NN for a dry canopy (in  $\mu g \, N \, m^{-2} \, s^{-1}$ ).

of  $\mathrm{NH_4^+}$  (69%) and was slightly higher during the summer period. Dry deposition was estimated at  $-30.7~\mathrm{kg}~\mathrm{N}~\mathrm{ha}^{-1}~\mathrm{yr}^{-1}$  (64% of TD) and consisted mainly of  $\mathrm{NH_x}$  ( $-26~\mathrm{kg}~\mathrm{N}~\mathrm{ha}^{-1}~\mathrm{yr}^{-1}$ ). Dry deposition of  $\mathrm{NH_3}$  contributed for 41% and 64% to TD and DD, respectively, and its share was even higher during the summer period when its amount raised to  $-29.3~\mathrm{kg}~\mathrm{ha}^{-1}$  on annual basis. Due to its preponderance in WD and DD,  $\mathrm{NH_x}$  contributed for 79% to TD. Share of  $\mathrm{NO_y}$  compounds in DD was minor compared to  $\mathrm{NH_x}$  and was of similar magnitude as  $\mathrm{WD}_{\mathrm{NO_7}}$ .

Throughfall (TF) of  $NH_4^+$  amounted to  $-23 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  and was 50% higher during the summer period (Table 6). This was in line with higher estimated DD of  $NH_3$  during the summer period. Total deposition of  $NH_x$  exceeded  $TF_{NH_4^+}$  with  $-15 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  (or 31% of TD) indicating substantial uptake of  $NH_x$  by the canopy. Canopy uptake of  $NH_x$  was higher during the summer period with canopy uptake rates being more than twice as large compared to winter period.

Good agreement was observed when the sum of WD of  $\mathrm{NH}_4^+$  and DD fluxes of  $\mathrm{NH}_3$ , which were accumulated — and partly washed off — during the 2—3 weeks exposure of the TF collectors, were regressed against the TF amounts captured during the same sampling interval (Fig. 7). Throughfall amounts of  $\mathrm{NH}_4^+$  (expressed per ha and per sampling event), which varied between 0.1 and 3.6 kg N ha<sup>-1</sup> sampling<sup>-1</sup>, were positively correlated (n=42;  $R^2=0.70$ ) with the sums of estimated DD of  $\mathrm{NH}_3$  and measured  $\mathrm{WD}_{\mathrm{NH}_4^+}$  captured within the same sampling interval. The slope of the regression line exceeded unity confirming presence of canopy uptake.

Throughfall of  $NO_3^-$  amounted to  $-10.1 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  and was also 50% higher during the summer period. This was, however, not in agreement with DD of  $NO_y$  the global estimate of which was more elevated during the winter. Throughfall of  $NO_3^-$  did not differ from TD of  $NO_y$  indicating lacking canopy uptake.

# 4. Discussion

# 4.1. Concentrations

Analysis of the nitrogen levels at a suburban forest near Antwerp revealed the presence of high gaseous  $NO_x$ . Nitrogen

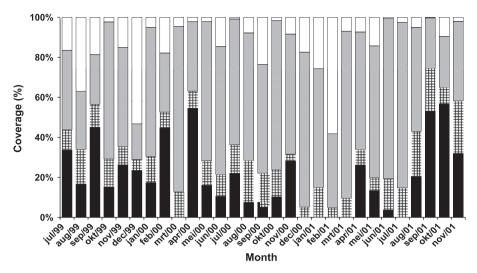


Fig. 5. Monthly data coverage over July 1999—November 2001 including: (1) data measured (in black); (2) data assumed zero (crossed); (3) data NN filled (in grey); (4) data not filled (in white).

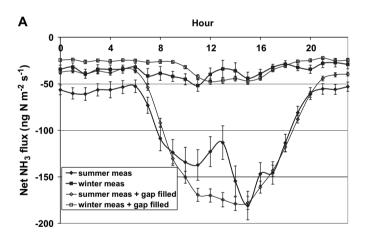
dioxide (NO<sub>2</sub>) was found to be the most important N compound, followed by NO and NH3. High pollution levels of NO and NO2 in southwesterly winds at the measuring site were due to vehicle emissions or other (industrial, suburban) combustion sources. The correlation between HONO and NO/NO<sub>2</sub> and their respective wind dependencies indicated a similar emission source, or its formation from these precursors. The ratio of HONO/NO<sub>2</sub> amounted to 8%, which is much higher than the ratio measured by nearby combustion sources (<1%) (Kurtenbach et al., 2001). On site formation of HONO from NO<sub>2</sub>, which could occur through reaction with aerosols (Reisinger, 2000), water vapour (Harrison et al., 1996) or light absorbing organic substances (Kleffmann et al., 2003), could therefore not be precluded. Although HONO is strongly susceptible to photolysis during the daytime, its levels were higher than for HNO<sub>3</sub>. The latter compound was found to be low in concentration, which was possibly due to the generally high ammonia background concentrations in Flanders.

High levels of reduced nitrogen originating from agricultural activities situated in the eastern sector contributed significantly to the nitrogen pollution climate. The presence of NH<sub>3</sub> also led to the advection of significant amounts of NH<sub>4</sub>- and NO<sub>3</sub>-bearing aerosols, converted from NH<sub>3</sub> during transport to the site. The variable distance in location of the emission sources (less than 5 km to more than 10 km) and differences in the vegetation composition in the upwind direction might be the reason for the observed variability in the NH<sub>3</sub>/NH<sub>4</sub> ratio of the air masses reaching the tower site for a given wind sector.

# 4.2. Fluxes

Dry deposition fluxes of nitrogen were dominated by  $NH_3$  for which the flux was estimated at -64 ng N m<sup>-2</sup> s<sup>-1</sup>. Fluxes of a similar magnitude (55–85 ng N m<sup>-2</sup> s<sup>-1</sup>) were measured in the Netherlands above a Douglas forest (Wyers et al., 1992; Wyers and Erisman, 1998). In order to complete the annual  $NH_3$  budget, gaps in data set were filled by estimates obtained by neural networks. Neural networks performed the

best when the canopy was dry, which was the predominant canopy wetness category (44% of all cases) at our site (Neirynck et al., 2005). During these conditions more turbulence is generated and fluxes can be measured (and estimated) more precisely. During rainy conditions, deviations from measured fluxes



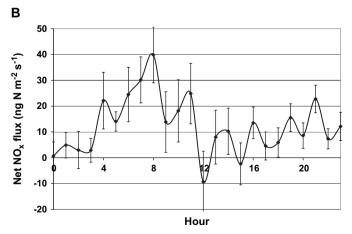


Fig. 6. Averaged diurnal course of measured (n = 8824) and gap filled (n = 37,123) NH<sub>3</sub> fluxes (A) and measured (n = 26,603) NO<sub>x</sub> fluxes (B) (in ng N m<sup>-2</sup> s<sup>-1</sup>). Standard errors are shown by bars.

Table 5 Averaged flux characteristics for N compounds measured on daily basis

_		_		
	Period	n	$F (\text{ng N m}^{-2} \text{ s}^{-1})^{\text{a}}$	$v_{\rm d}$ (cm s <sup>-1</sup> )
pNH <sub>4</sub>	Sep 99 till Jun 00	9309	-20	1.2
$pNO_3$	Jun/Sep/Oct 00	2360	-5.0	1.2
	Jan/Feb/Mar 01	2356	-16	1.5
$HNO_3$	Jun/Sep/Oct 00	3317	-6.4	4.2
	Jan/Feb/Mar 01	3040	-2.0	4.5
HONO	Jun/Sep/Oct 00	3317	-8.6	2.4
	Jan/Feb/Mar 01	3040	-9.2	2.0

 $<sup>^{\</sup>rm a}$  Fluxes were obtained by multiplying the half hourly-modelled  $\nu_{\rm d}$  with daily measured concentrations.

were the worst. In these conditions, which represent 9% of the data set, fluxes are hard to measure and so were simulations.

The flux of  $NH_3$  exhibited a strong seasonal variability, with summer deposition three times higher compared to winter conditions (Table 2, Fig. 6A). This is due to higher  $NH_3$  levels as a consequence of higher temperatures during summer and manure spreading prohibition during winter time. Deposition velocities did not vary substantially across the season although a decreased  $\nu_d$  was noticed during summer at night (Table 3).

When NO<sub>x</sub> fluxes were calculated with the gradient method, the net flux appeared to be upward (8.5 ng N m<sup>-2</sup> s<sup>-1</sup>), with the highest net-emission of NO<sub>x</sub> measured in the morning (40 ng N m<sup>-2</sup> s<sup>-1</sup>). Above-canopy fluxes calculated using the gradient analysis incorporate, besides deposition from ambient NO<sub>x</sub>, also emitted NO<sub>x</sub> fluxes from the canopy surface. We speculate that NO<sub>x</sub> emissions originated from soil-emitted NO, typically occurring in nitrogen-impacted forests (http:// 195.127.136.75/nofretete/). Emitted NO was probably stored in the trunk or canopy space during stable events, as indicated by the distinct morning peak (Fig. 2). Upon the initiation of turbulence, the stored NO below the canopy underlies rapid oxidation to NO<sub>2</sub> in the presence of ozone entrained from above (Dorsey et al., 2004; Duyzer et al., 2004). The surplus of NO<sub>2</sub> formed above the canopy led to negative gradients and upward fluxes, especially during the morning. Later at noon, the net NO<sub>x</sub> flux tended to be directed towards the canopy, possibly as result of increased deposition of airborne NO<sub>2</sub>. Such a dual pattern of counteracting fluxes was also observed in a Douglas fir forest in the Netherlands (Duyzer et al., 2004).

Fluxes of pNH<sub>4</sub> and pNO<sub>3</sub> have been estimated using parameterisations for a Douglas fir stand in the Netherlands. The application of these models was justified since deposition velocities of small aerosols were found to be underestimated for very rough surfaces (Gallagher et al., 1997; Ruijgrok et al., 1997). Modelled  $\nu_{\rm d}$  of pNH<sub>4</sub> (1.2 cm s<sup>-1</sup>) were similar to those measured for ammonium(bi)sulphate (MMD 0.8 µm) by Wyers et al. (1995) over the forest site in the Netherlands, with average values of 1.2–1.5 cm s<sup>-1</sup>. They were, however, higher than experimentally derived  $\nu_{\rm d}$  of pNH<sub>4</sub> for ammonium sulphate particles in Hungary (0.8 cm s<sup>-1</sup>) (Horváth, 2003) or when obtained with wash-off methods (Marques et al., 2001 and literature mentioned therein).

Wet deposition (WD) and dry deposition (DD) of NH<sub>s</sub> and NO<sub>s</sub> versus annual throughfall (TF) of NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> (in kg N ha<sup>-1</sup> yr<sup>-1</sup>)

		$NH_x$					NOy				$NH_x + NO_y$			
		Summer	Winter	Total	% TD		Summer	Winter	Total	%TD	Summer	Winter	Tot	%TD
WD	NH <sup>+</sup>	-13.5	-10.5	-12.0	25%	$NO_3^-$	-5.7	-5.2	-5.5	11%	-19.3	-15.7	-17.5	36%
DD	$NH_3$	-29.3	-9.8	-19.6	41%	$NO_x$	2.5	3.0	2.7	<i>2</i> /29—				
	$pNH_4$	-6.2	-6.8	-6.5	13%	ONOH	-2.7	-2.9	-2.8	%9				
						$HNO_3$	-2.0	9.0-	-1.3	3%				
						$pNO_3$	-1.6	-5.0	-3.3	2%				
	Sum	-35.5	-16.5	-26.0	54%	Sum	-3.8	-5.6	7.4-	%0I	-39.3	-22.1	-30.7	64%
TD	NHx	-49.1	-27.0	-38.0	2662	NO	9.6-	-10.8	-10.2	21%	-58.6	-37.8	-48.2	%00I
TF	NH <sup>+</sup>	-27.9	-18.1	-23.0	48%	$NO_3^-$	-12.1	-8.1	-10.1	21%	-40.0	-26.1	-33.1	%69
Canopy	uptake <sup>a</sup>	-21.1	-8.9	-15.0	31%		2.5	-2.7	-0.1	%0	-18.6	-11.7	-15.2	31%

Percent contribution of each compound in total nitrogen deposition is given in additional column.

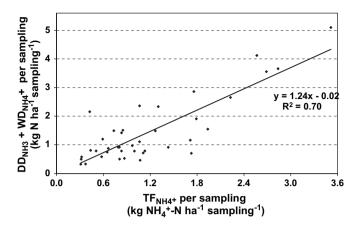


Fig. 7. Sum of estimated dry deposition of  $NH_3$  ( $DD_{NH_3}$ ) and measured wet deposition of  $NH_4^+$  ( $WD_{NH_4^+}$ ) captured per sampling interval of 2–3 weeks, plotted against measured throughfall  $NH_4^+$  ( $TF_{NH_4^+}$ ) measured during the same sampling interval.

Modelled  $\nu_{\rm d}$  for HNO<sub>3</sub> were in line with those determined for deciduous forest (Meyers et al., 1989; Pryor et al., 2002) but lower than for coniferous forests (Pryor and Klemm, 2004; Sievering et al., 2001). Fluxes and concentrations differed strongly between the two measuring seasons, which might reflect a shift in the thermodynamic equilibrium between gaseous HNO<sub>3</sub> and pNO<sub>3</sub>. According to Seinfeld and Pandis (1998), higher temperatures (which we encountered during the summer/autumn campaign) favoured shifts in concentrations of HNO<sub>3</sub> relative to pNO<sub>3</sub>. This led to opposing seasonal deposition patterns of both compounds above a pine forest in Germany (Dämmgen and Zimmerling, 2002).

Deposition velocities of HONO were estimated to be lower than those for HNO<sub>3</sub>, but higher concentrations compensated, resulting in fluxes of a similar magnitude. The mean  $\nu_d$  of 2–2.4 cm s<sup>-1</sup> was similar to modelled values of Dämmgen and Zimmerling (2002) but higher than the modelled mean of 1.3 cm s<sup>-1</sup> for a spruce forest in Germany (Zimmermann et al., 2006).

### 4.3. Comparison of total deposition with throughfall data

When all compounds were added up, a TD of  $-48 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  was obtained, with  $\text{NH}_x$  being the main contributor to DD and WD. This estimate largely exceeded the nitrogen TF deposition of  $-33 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ , which is in line with other nitrogen flux studies on nitrogen-impacted forest sites (Dämmgen and Zimmerling, 2002; Marques et al., 2001; Schmitt et al., 2005). Although this discrepancy could partly be associated with general difficulties in relating throughfall measurements to the footprint of the measuring site, canopy uptake of N could not longer be ignored.

Uptake of N by the canopy is primarily the result of ion uptake by foliar and branch tissues or active uptake by canopy lichens and micro organisms (Lovett and Lindberg, 1993). Canopy budget models (Ulrich, 1983) are often applied to correct TF for possible canopy uptake of NH<sub>4</sub><sup>+</sup> exchanged for base cations. These models are quite senseless at our site since

possibilities for base cation exchange were negligible (Neirynck et al., 2004). Ferm and Hultberg (1999) suggested that dry-deposited  $NH_x$ , consumed by the canopy, could later reappear as organic N in the TF samplers. Recent measurements of organic N showed at our site that DON constituted a substantial fraction of the total nitrogen concentration (10–15% of  $N_{tot}$ ). Throughfall enrichment by organic N originating from deposited inorganic N, utilised at the canopy level, could therefore not be excluded.

# 4.4. Possible uncertainties in nitrogen fluxes and nitrogen budget

Part of the flux estimates relies on resistance models, whose uncertainty is difficult to quantify. The advantages of such models include simplicity, and widespread use allowing for comparison with other sites. In the case of particle deposition, the momentum analogy employed (Eq. (10)) belies differences in canopy sink mechanisms for momentum and particles: inertial impaction of particles is most efficient for the smallest obstacles, which in the canopy would be needles (Langmuir, 1961) but obstacle size has not been taken into account in the collection efficiency *E.* Momentum, on the other hand, is more efficiently absorbed by form (pressure) drag over more bulky protrusions (Thom, 1975).

The fact that only small-size particles (PM<sub>2.5</sub>) were collected and analysed in our measuring programme might cast further doubt on our particle deposition results. This might constitute an underestimate, especially for pNO<sub>3</sub>, which include particles over a wide size range (Lovett and Lindberg, 1986). At coastal areas, reactions between marine dominated and polluted air masses may shift pNO<sub>3</sub> onto the coarse mode (Yeatman et al., 2001). Schaap et al. (2002) though, provided strong arguments that nitrate concentrations over continental Western Europe were predominantly present in the fine aerosol fraction.

In this study, nitrogen compounds were measured at two different time scales, and daily sampling of some nitrogen compounds was required to ensure high analytical precision. The short-term variation was therefore not completely captured and, depending on the relation between concentration and deposition velocity, a non-systematic bias could be expected (Andersen and Hovmand, 1999).

The fact that fluxes are assumed to be proportional to concentrations when using the resistance approach gives rise to further criticism of results. In case of HONO and HNO<sub>3</sub>, it was assumed that there was no substantial surface concentration and no emission could occur. Consequently,  $\nu_{\rm d}$  and  $R_{\rm c}$  are independent of measured concentration and fluxes can simply be derived from  $\nu_{\rm d}$ . For some gases, however, a surface concentration or compensation point must be considered, because of limited uptake capacity in certain conditions and occurrence of emissions. In our measuring conditions,  $\nu_{\rm d}$  of NH<sub>3</sub> was found to decrease with increasing ammonia levels because of saturation effects (Neirynck et al., 2005). Non-linearity between flux and concentration was also noticed by Horii et al. (2004) for NO<sub>2</sub> and Tarnay et al. (2002) for HNO<sub>3</sub>.

The bidirectional nature of some species (NH<sub>3</sub>, NO<sub>x</sub>, HONO) obviously thwarts the unambiguous completion of the nitrogen budget at our site. At present, we can only conjecture what the sources and sinks for these species might be and in-depth analysis is required. In case of NH<sub>3</sub>, emission events were observed and both stomata as saturated water films were surmised to be the cause of these emissions (Neirynck et al., 2005). Bidirectional models should therefore be preferred for gap filling procedures of NH3 fluxes instead of neural networks although complex compensation point models require much information about stomatal compensation points along with acidity and thickness of water films covering the leaves (Sutton et al., 1998). Also more information is required about the chemical behaviour of HONO, which was assumed to act similarly to SO<sub>2</sub>. We presume that the estimate of the HONO flux is associated with large uncertainties since a proportion of HONO measured above the canopy might be released from the forest floor after reduction of NO<sub>2</sub> and also subjected to emission (Kleffmann et al., 2003). Inclusion of downward HONO flux in the nitrogen budget may therefore overestimate the real N input. Also sources and sinks for NO<sub>x</sub> have not yet been clearly identified at our site. More information is needed about the impact of possible NO soil emission on the budget of NO<sub>v</sub>. Part of the NO converted to NO<sub>2</sub> can also be recovered by the canopy through stomatal uptake (Duyzer et al., 2004), further entangling the nitrogen budget.

# 5. Conclusions

Nitrogen dioxide was found to be the main nitrogen compound measured above this nitrogen-saturated suburban forest. It was shown that the canopy surface acted as a source for  $NO_x$  with highest emissions in the morning.

Reduced nitrogen was found to be the main contributor to dry and wet deposition. Total deposition at our site was ruled predominantly by dry deposition, given the neighbouring presence of agricultural, combustion and industrial sources. Sites located further away from these sources will be less exposed because dry-deposited substances are depleted from the air travelling downwind (Lovett, 1994). Such environmental conditions are, however, rarely met in a densely populated and industrially developed region such as Flanders.

Total nitrogen exceeded throughfall inputs, which was in line with other nitrogen flux studies at nitrogen-impacted sites. Although modelled dry deposition in these studies must sometimes be regarded as an upper limit, throughfall inputs systematically underestimate total inputs and hence, presence of substantial canopy sinks can therefore not be ignored (Lovett and Lindberg, 1993).

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